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## MAGNESIUM OXIDE-BASED SORBENTS FOR CARBON DIOXIDE CAPTURE: FROM FUNDAMENTALS TO PRACTICE

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Despite numerous efforts to decarbonize the energy sector, the rapid increase in CO<sub>2</sub> global emissions continues. Thus, the role of carbon capture and utilization in attaining net zero emissions remains indispensable. To further achieve so, several emerging technologies are being studied. This presentation aims to summarize various emerging sorbents and technologies for CO<sub>2</sub> capture. Among sorbents available for CO<sub>2</sub> capture, alkali nitrate saltpromoted MgO-based sorbents are given focus due to their high sorption capacities. In-situ TEM studies show that eutectic mixture (EM)-MgO sorbents undergo rearrangement and EM migration after multiple use, resulting in poor cyclability and reusability. Several strategies such as the use of stabilizers, supports, and employing a core-shell morphology were thus employed to achieve stable cyclic performances. Sorption mechanisms for each studied sorbent were also derived. Theoretical modeling suggests that an electric field (EF) promotes the adsorption of CO<sub>2</sub> on various CO<sub>2</sub> capture sorbents. However, this phenomenon has never been experimentally confirmed due to the lack of EF-responsive sorbents with measurable CO<sub>2</sub> sorption capacity. This study investigates the effect of an applied electric field on a eutectic mixture (EM)-promoted magnesium oxide sorbent with high CO<sub>2</sub> sorption capacity. Results show that the electric field effectively increased the CO<sub>2</sub> uptake of EM-MgO at significantly lower temperatures than its known operating range. Further investigation reveals that the electric field enhances the adsorption strength of CO<sub>2</sub> on the sorbent while also initiating the electrophoretic migration of EM ions, thus effectively redistributing the EM on the MgO surface during CO<sub>2</sub> sorption. The sample also exhibited stable sorption under cycling with alternating electric field directions, thus resolving the capacity loss commonly experienced by alkali nitrate salt-promoted MgO sorbents.

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